- (6) E. Baer, Biochem. Prep., 2, 31 (1952).
- (7) E. Baer and H. O. L. Fischer, J. Biol. Chem., 128, 463 (1969).
 (8) J. C. Sowden and H. O. L. Fischer, J. Am. Chem. Soc., 64, 1291
- (8) (1942).

- (1972).
 (9) B. Belleau and J. Puranen, J. Med. Chem., 6, 325 (1963).
 (10) D. Triggle and B. Belleau, Can. J. Chem., 40, 120 (1962).
 (11) R. E. Reeves, Adv. Carbohydr. Chem., 6, 107 (1951).
 (12) S. T. K. Bukhari, R. D. Gutherie, A. I. Scott, and A. D. Wrixon, Chem. Commun. 150 (1969). (12) S. H. K. Bukhari, H. D. Gutherie, A. I. Scott, and A. D. Wrixon, *Tetrahedron*,
 (13) S. T. K. Bukhari, R. D. Gutherie, A. I. Scott, and A. D. Wrixon, *Tetrahedron*,
- 26, 3653 (1970).
- (14) S. T. K. Bukhari and R. D. Gutherie, *Carbohydr. Res.*, **12**, 469 (1970).
 (15) S. T. K. Bukhari and R. D. Gutherie, *J. Chem. Soc. C*, 1073 (1969).
 (16) R. S. Downing and F. L. Urbach, *J. Am. Chem. Soc.*, **90**, 5344 (1968)
- (17) A. K. Ganguly, O. Z. Sarre, and J. Morton, Chem. Commun., 1488
- (1969).
- A. K. Mallams, J. Am. Chem. Soc., 91, 4505 (1969).
 R. W. Gillard and R. Wootton, J. Chem. Soc. B, 921 (1968).
 A. I. Scott and A. D. Wrixon, Chem. Commun., 1184 (1969).
- (21) A. Harner and C. S. P. Jenkins, *Chem. Commun.*, 350 (1969)
 (22) L. A. Mitscher, M. S. Bathala, and G. W. Clark, in preparation
- (23) L. A. Mitscher, P. W. Howison, and T. D. Sokowlowski, J. Med. Chem., 16,

- 98 (1973).
- (24) L. A. Mitscher, M. S. Bathala, G. W. Clark, P. W. Howison, and T. D. Sokowlowski, in preparation. J. Dillon and K. Nakanishi, *J. Am. Chem. Soc.*, **96**, 4057 (1974).
- (25)
- (26)K. Nakanishi and J. Dillon, J. Am. Chem. Soc., 93, 4098 (1971).
- (27) K. Nakanishi, D. A. Schoolery, M. Koreeda, and J. Dillon, Chem. Commun., 235 (1971).
- J. Dillon and K. Nakanishi, J. Am. Chem. Soc., 96, 5059 (1974).
 J. Dillon and K. Nakanishi, J. Am. Chem. Soc., 97, 5409 (1975).
 J. Dillon and K. Nakanishi, J. Am. Chem. Soc., 97, 5417 (1975).
- N. Harada and K. Nakanishi, *Acc. Chem. Res.*, **5**, 257 (1972), N. Harada and K. Nakanishi, *J. Am. Chem. Soc.*, **91**, 3989 (1969). (31)
- (32)
- (33) S. L. Chen, N. Harada, and K. Nakanishi, J. Am. Chem. Soc., 96, 7352 (1974).
- (34) E. Bunnenberg and C. Djerassi, J. Am. Chem. Soc., 82, 5953 (1960). (35)
- A. H. Haines and C. S. P. Jenkins, *Chem. Commun.*, 350 (1969). W. L. Nelson and C. E. Wood, Jr., *J. Chem. Soc., Chem. Commun.*, 896 (36)
 - (1973)
- (37) R. D. Gillard and H. M. Irving, Chem. Rev., 65, 603 (1965).
- (a) R. S. Cahn, C. K. Ingold, and V. Prelog, *Experentia*, **12**, 81 (1956); (b) *Angew. Chem., Int. Ed. Engl.*, **5**, 385 (1966).
 (39) J. C. Danilewicz and J. E. G. Kemp, *J. Med. Chem.*, **16**, 168 (1973).
- Synthesis and Reactions of 7-Hydrazonocephalosporanates

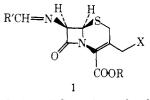
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p-Nitrobenzyl 7-hydrazonocephalosporanates 3 (R₁ = p-NO₂PhCH₂; R₂ = H) were synthesized and identified as isomers. Thienylacetylation and reduction gave the hydrazino compound 5. Compounds 3 react with NBS in aqueous acetone to give ketones 6. Reduction of 6 gives alcohol 7 ($R_1 = CH_2Ph$; $R_3 = H$) which was acylated and deblocked to give a series of cephalosporin oxygen analogues.

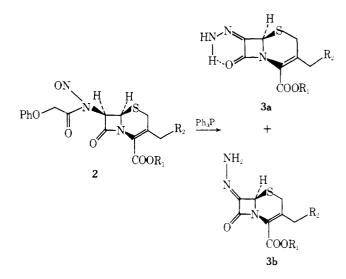
Many chemical modifications at C7 of cephalosporins have been achieved through activation of the C₇ position in such structures as 1. Another method of entry into this posi-



tion involves oxidation of C7 to form the diazo compound followed by further reactions characteristic of this group.¹

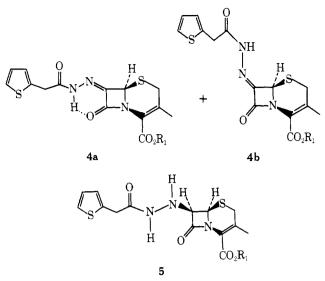
7-Diazocephalosporanates have been synthesized via diazotization of the amine,^{2,3} and rearrangement of 6β -N-nitrosophenoxyacetamidocephalosporanates in the presence of base.⁴ The latter reaction gives a poor yield since the N-nitroso amide is surprisingly resistant to rearrangement. Several new methods have been applied to this system and will be reported here.

The nitroso compound, p-nitrobenzyl 7β -N-nitrosophenoxyacetamidodeacetoxycephalosporanate 2 ($R_1 = p$ - NO_2PhCH_2 ; $R_2 = H$), reacts with triphenylphosphine to give a mixture of p-nitrobenzyl 7-hydrazonocephalosporanates 3 $(R_1 = p - NO_2 PhCH_2; R_2 = H)$. It is postulated that the triphenylphosphine forces the N-nitroso-diazo rearrangement, giving the diazo derivative, which forms an adduct generating 3 on hydrolysis. Phenoxyacetic acid was isolated as a byproduct. The trichloroethyl ester of 3 ($R_1 = CCl_3CH_2$; $R_2 =$ H) has been reported derived from the diazotization of 7aminocephalosporanate with isoamyl nitrite in formic acid.⁵ The analogous hydrazono compounds have also been synthesized in the penicillin series.⁶ However, the existence of two isomers has not been reported in either series. The isomers are separable by chromatography and can be distinguished by



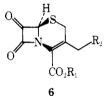
their physical properties. The intramolecular hydrogen bonding of structure 3a is expected to produce a less polar and lower melting compound. In addition the infrared absorption of the β -lactam carbonyl should be lowered by this bonding effect by about $20 \text{ cm}^{-1.7}$ These effects are observed and the structure assignments made accordingly. The isomers are interconvertible in the presence of base. Starting with either isomer, a mixture of both is obtained in the presence of pyridine.

Thienylacetylation of **3a** or **3b** ($R_1 = p \cdot NO_2PhCH_2$; $R_2 =$ H) gave a pair of isomers 4 ($R_1 = p$ -NO₂PhCH₂). Stereospecific reduction of the hydrazones with potassium borohydride and removal of the blocking group gave one product from both isomers, the hydrazino analogue of deacetoxycephalothin 5 $(\mathbf{R}_1 = \mathbf{H})$. The phenoxyacetyl, acetyl, and free hydrazino an-



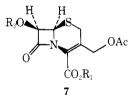
alogues have been reported 5 but without experimental details.

The hydrazono compound 3 ($R_1 = p$ -NO₂PhCH₂ or Ph₂CH; $R_2 = H$ or OAc) reacts with N-bromosuccinimide in aqueous acetone to give the ketone 6. The same compound can be synthesized from 7-diazocephalosporanates² by treatment with N-bromosuccinimide or N-bromoacetamide in aqueous acetone. This method is applicable to cephalosporin C to produce esters of 6. N,N-Phthaloylcephalosporin C di-



benzhydryl ester was treated with dinitrogen tetroxide and triphenylphosphine to give **3a** and **3b** ($R_1 = Ph_2CH$; $R_2 = OAc$), which react further with N-bromosuccinimide to give **6** ($R_1 = Ph_2CH$; $R_2 = OAc$).

Ketone 6 ($R_1 = Ph_2CH$; $R_2 = OAc$) was reduced with potassium borohydride to give the alcohol 7 ($R_1 = Ph_2CH$; R_3)



= H). Spectral data indicate that only the cis isomer is formed. Removal of the protecting group gave the oxygen analogue of 7-ACA, 7 ($R_1 = R_3 = H$). Acylation prior to deblocking gave the oxygen analogues of cephalosporin 7 [$R_1 = H$; $R_3 =$ PhCH(NH₂)CO, PhCH₂SO₂, PhOCH₂CO, C₄H₃SCH₂CO].

Compounds 7 ($R_1 = H$) were tested for bioactivity with the following (minimum inhibitory concentration, $\mu g/ml$ against *Staph. aureus* A100): $R_3 = PhCH_2SO_2$, 12.5; PhOCH₂CO, 12.5; C₄H₃SCH₂CO, 12.5.

Experimental Section

General. Melting points were determined on a Fisher-Johns melting point apparatus. Elemental analyses were performed by Galbraith Microanalytical Laboratories, Knoxville, Tenn. IR spectra were recorded on a Perkin-Elmer 237 spectrophotometer. NMR spectra were taken on a Varian T-60 spectrometer and are reported in parts per million downfield from Me₄Si. Baker-flex silica gel 1B-F was used for thin layer chromatography.

p-Nitrobenzyl 7β -N-Nitrosophenoxyacetamidodeacetoxycephalosporanate 2 ($\mathbf{R}_1 = p$ -NO₂PhCH₂; $\mathbf{R}_2 = \mathbf{H}$). Dinitrogen tetroxide (7 g) was dissolved in 100 ml of methylene chloride. A solution of *p*-nitrobenzyl 7 β -phenoxyacetamidodeacetoxycephalosporinate in methylene chloride (30 ml) was added in 20 min with stirring at 0 °C to a mixture of anhydrous sodium acetate (7 g), dinitrogen tetroxide (50 ml of above solution), and methylene chloride (50 ml). The mixture was stirred at 0 °C for 1 h. Additional portions of dinitrogen tetroxide (30 ml, 20 ml) were added immediately after and 30 min after addition of the cephalosporin derivative. Excess dinitrogen tetroxide was consumed by adding saturated sodium bicarbonate. The aqueous phase was extracted with methylene chloride (Na₂SO₄), and evaporated to a yellow oil. Crystallization from acetone–petroleum ether gave 3.0 g of yellow solid, 81%: mp 120–121 °C dec; [α]²⁵D –25.2° (c 0.76, CHCl₃); IR (CH₂Cl₂) 1790, 1745, 1725, 1535, 1350, 1225 cm⁻¹; NMR (CDCl₃) δ 8.28–7.50 (q, 4 H), 7.35–6.83 (m, 5 H), 5.87 (d, J = 4.5 Hz, 1 H), 5.57 (s, 2 H), 5.33 (d, J = 4 Hz, 2 H), 5.00 (d, J = 4.5 Hz, 1 H), 3.62–2.75 (q, J = 16 Hz, 2 H), 2.36 (s, 3 H).

Anal. Calcd for $C_{23}H_{20}N_4SO_8$ (512.49): C, 53.90; H, 3.93; N, 10.93; S, 6.26. Found: C, 53.82; H, 3.86; N, 10.76; S, 6.40.

Benzhydryl 7 β -(**Benzhydryl-5**-*N*,*N*-**phthaloyl-5**-**aminoadipamido**)**cephalosporanate**. *N*,*N*-**Phthaloylcephalosporin** C dibenzhydryl ester was prepared according to published procedures:⁶ mp 165–167 °C (lit. 161–163 °C); IR (CDCl₃) 3410, 3330, 1780, 1740, 1730, 1685, 1510, 1385, 1230 cm⁻¹; NMR (CDCl₃) δ 7.80 (m, 4 H), 7.30 (m, 20 H), 6.95 (s, 1 H), 6.90 (s, 1 H), 6.45 and 6.30 (2 s, 1 H), 5.82 (q, 1 H), 5.20–4.62 (m, 4 H), 3.70–3.02 (q, *J* = 19 Hz, 2 H), 2.43–2.22 (m, 4 H), 2.05 (s, 3 H), 1.90–1.60 (m, 2 H).

Benzhydryl 7 β -N-Nitrosobenzhydryl-5-N,N-phthaloyl-5aminoadipamidocephalosporanate. The above compound (0.5 g) was treated with dinitrogen tetroxide in the same manner as 2. A yellow oil was obtained: IR (film) 1780, 1730, 1720, 1525, 1385, 1240 cm⁻¹; NMR (CDCl₃) δ 7.80 (m, 4 H), 7.30 (m, 20 H), 6.95 (s, 1 H), 6.90 (s, 1 H), 5.79 (d, J = 4.5 Hz, 1 H), 5.30–4.70 (m, 4 H), 3.30 (s, br, 4 H), 2.40 (br, 2 H), 1.95 (s, 3 H), 1.90 (br, 2 H).

N-Nitrosoamide-Hydrazone Transformations. N-Nitrosoamide **2** was refluxed with 1 equiv of triphenylphosphine in benzene for 45 min. The reaction mixture was cooled to room temperature and was stirred vigorously with excess water for 2 h. Methylene chloride was added and the organic layer was washed with 5% sodium bicarbonate solution and water, dried (Na_2SO_4) , and evaporated to a brown oil. Separation of products (3) was effected by chromatography on silicic acid eluted by a mixture of methylene chloride and ethyl ether.

p-Nitrobenzyl 7-Hydrazonodeacetoxycephalosporanate 3a ($\mathbf{R}_1 = \mathbf{p}$ -NO₂PhCH₂; $\mathbf{R}_2 = \mathbf{H}$). R_f 0.45 (1:10 Et₂O-CH₂Cl₂). The product was crystallized from chloroform-petroleum ether, 28%: mp 115-116 °C; IR (CH₂Cl₂) 3435, 3300, 1760, 1730, 1610, 1525, 1350 cm⁻¹; NMR (CDCl₃) δ 8.25-7.50 (q, 4 H), 6.80 (s, 2 H), 5.38 (s, 2 H), 5.17 (s, 1 H), 3.73-3.00 (q, J = 19 Hz, 2 H), 2.19 (s, 3 H); MS M⁺ m/e 362; high-resolution mass spectrum gave molecular formula C₁₅H₁₄N₄SO₅.

p-Nitrobenzyl 7-Hydrazonodeacetoxycephalosporanate 3b ($\mathbf{R}_1 = \mathbf{p}$ -NO₂PhCH₂; $\mathbf{R}_2 = \mathbf{H}$). R_f 0.20 (1:10 Et₂O-CH₂Cl₂). The product was crystallized from chloroform, 58%: mp 165–166 °C; IR (CH₂Cl₂) 3430, 3290, 1780, 1730, 1615, 1525, 1350 cm⁻¹; NMR (CDCl₃) δ 8.25–7.50 (q, 4 H), 6.10 (s, 2 H), 5.35 (s, 2 H), 5.30 (s, 1 H), 3.65–3.00 (q, J = 18 Hz, 2 H), 2.20 (s, 3 H); MS M⁺ m/e 362; high-resolution mass spectrum gave molecular formula C₁₅H₁₄N₄SO₅.

Anal. Calcd for C₁₅H₁₄N₄SO₅ (362.36): C, 49.72; H, 3.89; N, 15.46; S, 8.85. Found: C, 49.54; H, 3.90; N, 15.32; S, 8.93.

Benzhydryl 7-Hydrazonocephalosporanate 3a ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{CH}$; $\mathbf{R}_2 = \mathbf{OAc}$). R_f 0.44 (1:10 Et₂O-CH₂Cl₂); oily product, yield 20%; IR (CHCl₃) 3455, 3320, 1760, 1730, 1620, 1375, 1220 cm⁻¹; NMR (CDCl₃) δ 7.38 (s, 10 H), 7.00 (s, 1 H), 6.85 (s, 2 H), 5.25 (s, 1 H), 5.23-4.55 (q, J = 14 Hz, 2 H), 3.75-3.10 (q, J = 19 Hz, 2 H), 2.02 (s, 3 H).

Benzhydryl 7-Hydrazonocephalosporanate 3b ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{CH}$; $\mathbf{R}_2 = \mathbf{OAc}$). R_f 0.24 (1:10 Et₂O-CH₂Cl₂). The product was crystallized from benzene: yield 45%; mp 152–154 °C; IR (CDCl₃) 3430, 3310, 1780, 1730, 1230 cm⁻¹; NMR (CDCl₃) δ 7.40 (s, 10 H), 6.98 (s, 1 H), 6.25 (s, 2 H), 5.28 (s, 1 H), 5.08–4.55 (q, J = 14 Hz, 2 H), 3.72–3.05 (q, J = 18 Hz, 2 H), 2.02 (s, 3 H).

Isomerizations of the Geometrical Isomers 3a and 3b. Starting with a pure isomer (either 3a or 3b, $R_1 = p$ -NO₂PhCH₂; $R_2 = H$) in methylene chloride in the presence of 1.5 equiv of pyridine, after a month at room temperature, a mixture of both isomers (3a and 3b) results. The isomers were separated by chromatography and their amounts were determined. Thus from 1 g of 3b, 0.45 g of 3a and 0.45 g of 3b were isolated; from 76 mg of 3a, 40 mg of 3a and 25 mg of 3b were isolated.

2-Thienylacetylation of 3b. *p*-Nitrobenzyl 7-hydrazonodeacetoxycephalosporanate **3b** ($R_1 = p$ -NO₂PhCH₂; $R_2 = H$; 1 g, 2.76 mmol), pyridine (0.33 ml, 1.5 equiv), and 2-thienylacetyl chloride (0.72 g, 1.5 equiv) were dissolved in methylene chloride (60 ml) at 0 °C. The solution was stirred at room temperature for 18 h and then diluted with methylene chloride (100 ml). The organic phase was successively washed by cold, dilute (0.1 N) hydrochloric acid, cold 5% sodium bicarbonate solution, and ice water, dried (Na₂SO₄), and evaporated to a brown oil. Products **4a** (0.7 g, 52%) and **4b** (R₁ = p-NO₂PhCH₂; R₂ = H; 0.5 g, 37%) were isolated by column chromatography on silicic acid eluted with 1:10 Et₂O-CH₂Cl₂.

2-Thienylacetylation of 3a. p-Nitrobenzyl 7-hydrazonodeacetoxycephalosporanate 3a ($R_1 = p$ -NO₂PhCH₂; $R_2 = H$; 0.2 g) was treated with pyridine and 2-thienylacetyl chloride as in the case of 3b. After column chromatography, the two products isolated were identical with those isolated in 2-thienylacetylation of 3b based on IR, NMR, and TLC data. Yields of products isolated were 4a (0.11 g, 41%); 4b (0.07 g, 23%).

p-Nitrobenzyl 7-(2-Thienylacetyl)hydrazonodeacetoxycephalosporanate 4a. R_f 0.45 (1:10 Et₂O–CH₂Cl₂); IR (CH₂Cl₂) 3290, 1770, 1720, 1680, 1520 cm⁻¹; NMR (acetone- d_6) δ 8.35–7.70 (q, 4 H), 7.35 (m, 1 H), 7.00 (m, 2 H), 5.58 (s, 1 H), 5.50 (s, 2 H), 4.20 (s, br, 2 H), 4.02–3.28 (q, J = 19 Hz, 2 H), 2.97 (s, br, 1 H), 2.24 (s, 3 H).

Anal. Calcd for $C_{21}H_{18}N_4S_2O_6$ (486.53): C, 51.84; H, 3.73; N, 11.52; S, 13.18. Found: C, 51.81; H, 3.70; N, 11.50; S, 13.17.

p-Nitrobenzyl 7-(2-Thienylacetyl)hydrazonodeacetoxycephalosporanate 4b. R_f 0.19 (1:10 Et₂O–CH₂Cl₂); mp 178–179 °C; IR (CH₂Cl₂) 3300, 1785, 1725, 1670, 1525 cm⁻¹; NMR (acetone- d_6) δ 8.38–7.72 (q, 4 H), 7.37 (m, 1 H), 7.00 (m, 2 H), 5.72 (s, 1 H), 5.50 (s, 2 H), 4.28 (s, br, 2 H), 3.84–3.21 (q, J = 18 Hz, 2 H), 2.90 (br, 1 H), 2.28 (s, 3 H).

Anal. Calcd for $C_{21}H_{18}N_4S_2O_6$ (486.53): C, 51.84; H, 3.73; N, 11.52; S, 13.18. Found: C, 51.56; H, 3.69; N, 11.26; S, 12.93.

Borohydride Reduction of 4a. To a cooled, stirred solution of p-nitrobenzyl 7-(2-thienylacetyl)hydrazonodeacetoxycephalosporanate (4a, $R_1 = p$ -NO₂PhCH₂; $R_2 = H$; 0.65 g, 1.34 mmol) in tetrahydrofuran (20 ml) was added a cold solution of potassium borohydride (0.16 g, 2.2 molar equiv) in 50% aqueous THF (30 ml). After 3 min. 1 N hydrochloric acid was added to bring the pH of the solution to 2. The solution was diluted with water and extracted twice with methylene chloride. The combined extracts was washed once with 5% sodium bicarbonate solution and once with water, dried (Na₂SO₄), and evaporated to a yellow solid. Crystallization from chloroform gave white crystalline p-nitrobenzyl 7β -(2-thienylacetyl)hydrazinodeacetoxycephalosporanate 5 ($R_1 = p$ -NO₂PhCH₂), 0.50 g, 72%: mp 183–184 °C dec; IR (CH₂Cl₂) 3390, 3280, 1775, 1725, 1680, 1520 cm⁻ NMR (CDCl₃) δ 8.20 (d, 2 H), 7.58 (d over br s, 3 H), 7.20 (m, 1 H), 6.95 (m, 2 H), 5.32 (s, 2 H), 5.27 (br, 1 H), 4.98 (d, J = 4.2 Hz, 1 H), 4.75 (m, 2 H), 5.32 (s, 2 H), 5.27 (br, 1 H), 4.98 (d, J = 4.2 Hz, 1 H), 4.75 (m, 2 H), 5.32 (s, 2 H), 5.27 (br, 1 H), 4.98 (d, J = 4.2 Hz, 1 H), 4.75 (m, 2 H), 5.32 (s, 2 H), 5.27 (br, 1 H), 4.98 (d, J = 4.2 Hz, 1 H), 4.75 (m, 2 H), 5.27 (br, 1 H)1 H), 3.78 (s, 2 H), 3.60–2.98 (q, J = 18 Hz, 2 H), 2.20 (s, 3 H).

Anal. Calcd for C₂₁H₂₀N₄S₂Õ₆ (488.54): C, 51.63; H, 4.13; N, 11.47; S, 13.13. Found: C, 51.44; H, 4.17; N, 11.25; S, 12.97.

Borohydride Reduction of 4b. *p*-Nitrobenzyl 7-(2-thienyl-acetyl)hydrazonodeacetoxycephalosporanate (**4b**, $R_1 = p$ -NO₂PhCH₂; $R_2 = H$; 0.61 g) was treated with potassium borohydride in the same manner as in the reduction of **4a**. TLC (silica gel, 1:4 Et₂O-CH₂Cl₂) of the workup showed starting material (**4b**, R_1 0.50) and a product (R_1 0.17). Column chromatography gave 0.35 g of unreacted **4b** (55% recovery) and a white crystalline solid (0.12 g, 20%) which is identical with 5 ($R_1 = p$ -NO₂PhCH₂) based on melting point, IR, NMR, and TLC data.

 7β -(2-Thienylacetyl)hydrazinodeacetoxycephalosporanic Acid 5 ($\mathbf{R}_1 = \mathbf{H}$). *p*-Nitrobenzyl 7 β -(2-thienylacetyl)hydrazinodeacetoxycephalosporanate (5, $\mathbf{R}_1 = p$ -NO₂PhCH₂, 0.10 g) was dissolved in glacial acetic acid (20 ml) and hydrogenated (1 atm) at room temperature for 2 h in the presence of 10% Pd/C (0.4 g). The catalyst was removed by filtration and washed with glacial acetic acid. The solvent was partially removed at reduced pressure and then freeze-dried together with an excess of benzene, leaving a pale yellow solid, 60 mg: IR (KBr) 3500–2700, 1770, 1730–1650 cm⁻¹; NMR (Me₂SO- d_6) δ 9.83 (br, 1 H), 7.45 (m, 1 H), 7.05 (m, 2 H), 6.65 (br, 1 H), 5.10–4.80 (m, 2 H), 3.75-3.15 (m, 2 H), 2.10 (br s, 3 H).

p-Nitrobenzyl 7-Oxodeacetoxycephalosporanate 6 ($\mathbf{R}_1 = \mathbf{p}$ -NO₂PhCH₂; $\mathbf{R}_2 = \mathbf{H}$). p-Nitrobenzyl 7-hydrazonodeacetoxycephalosporanate (3b, 0.9 g, 2.49 mmol) was dissolved in 10% aqueous acetone (200 ml), and the solution was cooled in an ice bath. Pyridine (3.65 ml, 0.04 mol) and N-bromosuccinimide (0.97 g, 2.2 equiv) were added to the stirred solution. After 45 min the reaction mixture was diluted with methylene chloride and cold water. Extraction with cold methylene chloride was repeated three times. The organic layer was washed successively with cold hydrochloric acid (0.1 N), cold 5% sodium bicarbonate solution, and ice water, dried (Na₂SO₄), and evaporated to a yellow oil, 0.94 g. Chromatography on silicic acid

eluted with 1:10 Et₂O-CH₂Cl₂ gave one major fraction, R_f 0.36 (1:10 Et₂O-CH₂Cl₂); 0.65 g (75%); IR (film) 1825, 1780, 1725, 1515 cm⁻¹; NMR of the product 6 (R₁ = p-NO₂PhCH₂; R₂ = H) after chromatography gave complicated signals suspected to be the result of a mixture of the ketone and the hydrate of the ketone. The same sample was refluxed in benzene under a Dean-Stark trap for 16 h and the following NMR was obtained after removal of solvent at reduced pressure: NMR (CDCl₃) δ 8.25-7.50 (q, 4 H), 5.40 (s, 2 H), 5.30 (s, 1 H), 3.78-3.10 (q, J = 19 Hz, 2 H), 2.25 (s, 3 H).

Benzhydryl 7-Oxocephalosporanate 6 ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{CH}$; $\mathbf{R}_2 = \mathbf{OAc}$). Compound 3b ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{CH}$; $\mathbf{R}_2 = \mathbf{OAc}$) (0.5 g, 1.11 mmol) was treated with N-bromosuccinimide in the same way as 3b ($\mathbf{R}_1 = p$ -NO₂PhCH₂; $\mathbf{R}_2 = \mathbf{H}$). A yellow oil was obtained after chromatography: 0.24 g, 49%; R_f 0.40 (1:9 Et₂O-CH₂Cl₂); IR (film) 1825, 1785, 1735, 1385, 1240 cm⁻¹; NMR of the product after chromatography indicated partial hydration. The sample after refluxing in benzene gave the following spectrum: NMR (CDCl₃) δ 7.39 (s, 10 H), 7.00 (s, 1 H), 5.21 (s, 1 H), 5.10-4.65 (q, J = 13 Hz, 2 H), 3.80-3.15 (q, J = 18 Hz, 2 H), 2.01 (s, 3 H).

Benzhydryl 7β-Hydroxycephalosporanate 7 ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{CH}$; $\mathbf{R}_3 = \mathbf{H}$). Crude 6 ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{H}$; $\mathbf{R}_2 = \mathbf{OAc}$) (2.95 g) was dissolved in THF (150 ml) and cooled to 0 °C. Potassium borohydride (0.74 g, 13.7 mmol) in 1:1 THF-H₂O (150 ml) was added quickly. The reaction was quenched after 2 min by addition of 1 N HCl to pH 2. The solution was diluted with water and extracted with methylene chloride, and the organic layer was washed with bicarbonate solution and salt solution. Drying and evaporation gave a yellow oil which was chromatographed to give 1.2 g of solid. Recrystallization from benzene gave mp 122-123 °C; IR (CH₂Cl₂) 3540, 1785, 1735, 1225 cm⁻¹; NMR (CDCl₃) δ 2.04 (s, 3 H), 3.45 (d, 2 H), 3.90 (s, 1 H), 4.62-5.20 (m, $J_1 = 4.5, J_2 = 13$ Hz, 3 H), 5.29 (d, J = 4.5 Hz, 1 H), 7.00 (s, 1 H), 7.39 (s, 10 H).

7β-Hydroxycephalosporanic Acid 7 ($\mathbf{R}_1 = \mathbf{R}_3 = \mathbf{H}$). Compound 7 ($\mathbf{R}_1 = \mathbf{Ph}_2\mathbf{H}$; $\mathbf{R}_3 = \mathbf{H}$) (0.3 g, 0.68 mmol) was dissolved in trifluoroacetic acid (7 ml) and anisole (1 ml) at 0 °C. After 1 h the solvents were evaporated and the residual oil washed with petroleum ether. The oil was dissolved in ethyl acetate and decolorized with charcoal. Crystallization from ethyl acetate gave 0.17 g (99%): mp 132 °C dec; IR (KBr) 3430, 3100, 1780–1700, 1625, 1380, 1220 cm⁻¹; NMR (acetone-d₆) δ 2.04 (s, 3 H), 3.55 (d, 2 H), 4.80–5.15 (m, J = 4.8, 13 Hz, 3 H), 5.40 (d, J = 4.8 Hz, 1 H).

Benzhydryl 7 β -Phenoxyacetoxycephalosporanate 7 (R₁ = CHPh₂; R₃ = PhOCH₂CO). Compound 7 (R₁ = CHPh₂; R₃ = H) (0.8 g, 1.8 mmol) and phenoxyacetyl chloride (0.42 g, 1.5 equiv) were dissolved in CH₂Cl₂ (50 ml). Pyridine (0.15 ml, 1.5 equiv) was added to the cooled, stirred solution. After 3 h stirring at room temperature, the solution was washed with water, bicarbonate, and salt solution. The solution was dried and evaporated and the residue was chromatographed on silicic acid with Et₂O-CH₂Cl₂ (1:20) to give 0.85 g (88%) of an oil: IR (film) 1785, 1730, 1600, 1495, 1380, 1225 cm⁻¹; NMR (CDCl₃) δ 1.98 (s, 3 H), 3.38 (s, 2 H), 4.75 (s, 2 H), 4.65-5.20 (q, J = 4.8, 14 Hz, 3 H), 6.10 (d, J = 4.8 Hz, 1 H), 6.80-7.54 (m, 16 H).

7β-Phenoxacetoxycephalosporanic Acid **7** ($\mathbf{R}_1 = \mathbf{H}$; $\mathbf{R}_3 = \mathbf{PhOCH}_2\mathbf{CO}$). Compound **7** ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{PhOCH}_2\mathbf{CO}$) was deblocked in the same way as **7** ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{PhOCH}_2\mathbf{CO}$) was 1230 cm⁻¹; NMR (acetone- d_6) δ 2.02 (s, 3 H), 3.60 (d, 2 H), 4.92 (s, 2 H), 4.70-5.28 (q, J = 14 Hz, 2 H), 5.25 (d, J = 4.8 Hz, 1 H), 6.32 (d, J = 4.8 Hz, 1 H), 6.87-7.42 (m, 5 H), 8.10 (s, 1 H).

Benzhydryl 7β -(2-Thienyl)acetoxycephalosporanate 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{C}_4\mathbf{H}_3\mathbf{SCH}_2\mathbf{CO}$). Compound 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{H}$) (0.45 g, 1.0 mmol), 2-thienylacetic acid (0.21 g, 1.5 equiv), and pyridine (0.1 ml, 1.2 equiv) were dissolved in $\mathbf{CH}_2\mathbf{Cl}_2$ (50 ml) at 0 °C. Disopropylcarbodiimide (0.13 g, 1 equiv) was added and the solution stirred at 0 °C for 1 h and stored at 5 °C for 17 h. The solution was filtered, diluted with $\mathbf{CH}_2\mathbf{Cl}_2$, and washed with cold dilute \mathbf{HCl} , bicarbonate, and salt solution. Drying and evaporation gave a yellow oil which was chromatographed on silicic acid with $\mathbf{Et}_2\mathbf{O}-\mathbf{CH}_2\mathbf{Cl}_2$ (1:20) to give 0.6 g (95%): IR (film) 1785, 1730, 1360, 1235 cm⁻¹; NMR (CDCl₃) δ 1.98 (s, 3 H), 3.36 (s, 2 H), 3.91 (s, 2 H), 4.60-5.20 (d on q, J = 4.8, 14 Hz, 3 H), 6.05 (d, J = 4.8 Hz, 1 H), 6.90–7.54 (m, 14 H).

7β-(2-Thienyl)acetoxycephalosporanic Acid 7 ($\mathbf{R}_1 = \mathbf{H}; \mathbf{R}_3 = \mathbf{C}_4\mathbf{H}_3\mathbf{SCH}_2\mathbf{CO}$). Compound 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{C}_4\mathbf{H}_3\mathbf{SCH}_2\mathbf{CO}$) was deblocked as described for 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{PhOCH}_2\mathbf{CO}$). Freeze-drying from benzene gave 98%: IR (film) 3560–2540, 1780, 1725, 1380, 1225 cm⁻¹; NMR (CDCl₃) δ 2.13 (s, 3 H), 3.47 (s, 2 H), 4.00 (s, 2 H), 4.82–5.38 (d on q, J = 4.8, 15 Hz, 3 H), 6.19 (d, J = 4.8 Hz, 1 H), 7.00 (d, 1 H), 7.20–7.40 (m, 2 H), 7.73 (s, 1 H).

Benzhydryl 7β -**Benzylsulfonylcephalosporanate** 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{PhCH}_2\mathbf{SO}_2$). Compound 7 ($\mathbf{R}_1 = \mathbf{CHPh}_2$; $\mathbf{R}_3 = \mathbf{H}$) was

benzylsulfonated with benzylsulfonyl chloride and pyridine as described for 7 ($R_1 = CHPh_2$; $R_3 = PhOCH_2CO$). The product was chromatographed on silica gel with Et₂O-CH₂Cl₂ (1:20) to give 74% of an oil: IR (film) 1780, 1725, 1625, 1495, 1450, 1370, 1220 cm⁻¹; NMR (CDCl₃) & 2.05 (s, 3 H), 3.45 (s, 2 H), 4.58 (s, 2 H), 4.55-5.20 (d on q, J = 13, 21 Hz, 3 H), 5.70 (d, J = 4.5 Hz, 1 H), 6.95 (s, 1 H), 7.45 (m, 15 H).

 7β -Benzylsulfonylcephalosporanic Acid 7 ($\mathbf{R}_1 = \mathbf{H}; \mathbf{R}_3 =$ **PhCH**₂**SO**₂). Compound 7 ($R_1 = CHPh_2$; $R_3 = PhCH_2SO_2$) was deblocked as described for 7 ($R_1 = CHPh_2$; $R_3 = PhOCH_2CO$) to give a yellow oil. Treatment with potassium 2-ethyl hexanoate gave 60 mg of the potassium salt: IR (KBr) 2910, 1755, 1725, 1600, 1360, 1225 $\rm cm^{-1}$

Benzhydryl 7ß-(2-tert-Butoxycarbonylamino-D-phenylacetoxy)cephalosporanate 7 [$R_1 = CHPh_2$; $R_3 = PhCH(NHCO_2$ t-Bu)CO]. Compound 7 ($R_1 = CHPh_2$; $R_3 = H$) (2.8 g, 6.4 mmol) was esterified with 2-tert-butoxycarbonylamino-D-phenylacetic acid as described for 7 ($R_1 = CHPh_2$; $R_3 = C_4H_3SCH_2CO$). Chromatography on silicic acid with $Et_2O-CH_2Cl_2$ (1:20) gave 0.5 g (11%) of oil: IR (film) 3300, 2960, 1790, 1720, 1500 cm⁻¹; NMR (CDCl₃) δ 1.90 (s, 3 H), 4.06 $(q, J_{gem} = 15 \text{ Hz}, 2 \text{ H}), 4.82 \text{ (d}, J = 4 \text{ Hz}, 2 \text{ H}), 5.35 \text{ (d}, J = 13 \text{ Hz}, 1 \text{ H}),$ 5.60 (m, 2 H), 6.03 (d, J = 4.8 Hz, 1 H), 6.86 (s, 1 H), 7.24 (m, 15 H).

 7β -(2-Amino-D-phenylacetoxy)cephalosporanic Acid 7 [\mathbf{R}_1 = H: \mathbf{R}_3 = PhCH(NH₂)CO]. Compound 7 [\mathbf{R}_1 = CHPh₂; \mathbf{R}_3 = PhCH(NHCO₂-t-Bu)CO] (0.13 g, 0.149 mmole) was deblocked as described for 7 ($R_1 = CHPh_2$; $R_3 = PhOCH_2CO$). After evaporation of the solvents the residue was dissolved in 10 ml of cold dioxane and 20 ml of cold methylene chloride. Toluenesulfonic acid (29 mg) was added and the solution freeze-dried. The residue was crystallized from dioxane-ether to give a white solid 7 $[R_1 = H; R_3 = D-PhCH(NH_3+CH_3PhSO_3-)CO]$: mp 138-140 °C; IR (KBr) 3400, 2900, 1760, 1730, 1610, 1500, 1375 cm⁻¹; NMR (acetone- d_6) δ 1.80 (s), 2.13 (s, 3 H), 3.12 (m, 2 H), 4.70 (d, J = 4 Hz, 2 H), 4.90 (d, J = 5 Hz, 1 H), 5.33 (s, 1 H), 6.15 (d, J = 4 Hz, 1 H), 6.90-7.50 (m, 9 H).

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Registry No.—2 ($R_1 = p$ -NO₂PhCH₂; $R_2 = H$), 51056-21-4; 3 (R_1 = p-NO₂PhCH₂; R₂ = H), 61394-33-0; 3 (R₁ = Ph₂CH; R₂ = OAc), 61394-34-1; 4 (R₁ = p-NO₂PhCH₂; R₂ = H), 61394-35-2; 5 (R₁ = p- NO_2PhCH_2), 61394-36-3; 5 (R₁ = H), 61394-37-4; 6 (R₁ = p- NO_2PhCH_2 ; $R_2 = H$), 61394-38-5; 6 ($R_1 = Ph_2CH$; $R_2 = OAc$), 59128-58-4; 7 ($R_1 = CHPh_2$; $R_3 = PhCH_2SO_2$), 61394-39-6; 7 ($R_1 = CHPh_2$; $R_3 = PhCH_2SO_2$), 61394-39-6; 7 ($R_1 = CHPh_2$; $R_2 = PhCH_2SO_2$), 61394-39-6; 7 ($R_2 = PhCH_2SO_2$), 6140-39-6; 7 (R_2 = PhCH_2SO_2), 6 PhCHNH₂COMeC₆H₄SO₃H), 61394-42-1; dinitrogen tetroxide, $10544-72-\overline{6}$; *p*-nitrobenzyl 7^β-phenoxyacetamidodeacetoxycephalosporanate, 28974-31-4; benzhydryl 7β -(benzhydryl-5-N,Nphthaloyl-5-aminoadipamido)cephalosporanate, 16361 - 81 - 2: benzhydryl 7β-N-nitrosobenzhydryl-5-N,N-phthaloyl-5-aminoadipamidocephalosporanate, 61394-43-2; 2-thienylacetyl chloride, 50529-60-7; phenoxyacetyl chloride, 701-99-5; 2-thienylacetic acid, 1918-77-0; benzylsulfonyl chloride, 1939-99-7; 2-tert-butoxycarbonylamino-D-phenylacetic acid, 33125-05-2.

References and Notes

- For review see P. G. Sammes, *Chem. Rev.*, **76**, 113 (1976).
 L. D. Cama, R. J. Leanza, T. R. Beattie, and B. G. Christensen, *J. Am. Chem.*
- L. D. Gama, H. J. Leanza, T. H. Bearle, and D. G. Grindenson, J. A. B. Soc., 94, 1408 (1972).
 J. S. Wiering and H. Wynberg, J. Org. Chem., 41, 1574 (1976).
 J. C. Sheehan, Y. S. Lo, J. Löliger, and C. C. Podewell, J. Org. Chem., 39, 1444 (1974); D. Hauser and H. P. Sigg, Helv. Chim. Acta, 50, 1327 (1967).
- (5) G. Lowe, German Patent 2 305 972 (1973); Chem. Abstr., 79, 115602 (1973).
 (6) D. M. Brunwin and G. Lowe, J. Chem. Soc., Chem. Commun., 192
- (1972).
 (7) F. Scheinmann, Ed., "An Introduction to Spectroscopic Methods for the
- Identification of Organic Compounds", Vol. I, Pergamon Press, Elmsford, N.Y. 1970, p 129.

1-Oxo-1,2,5-thiadiazolidin-3-ones. A Structural Reassignment

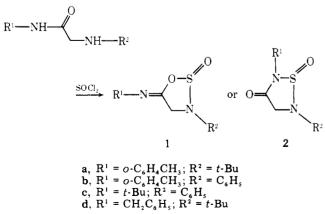
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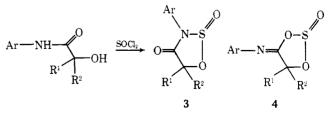
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The structure of the products obtained from the reactions of 2-aminoamides with thionyl chloride has been reinvestigated by means of ¹³C NMR spectroscopy. As a result of these investigations and comparisons with appropriate model compounds, the revised 1-oxo-1,2,5-thiadiazolidin-3-one structure has been assigned.

We reported previously the reaction of 2-aminoamides with thionyl chloride to produce 2-oxo-5-imino-1,2,3oxathiazolidines (1).¹ This structure was based on IR and



NMR spectral data as well as the mild acid hydrolysis of 1a to its precursor 2-aminoamide. Subsequently, Chupp reported on a similar reaction between 2-hydroxyarylamides and thionyl chloride.² Consideration of the IR and NMR spectral data led this author to prefer the 2-oxo-1,2,3-oxathiazolidin-4-one structure (3) over the isomeric 2-oxo-4-imino-1,3,2-dioxathiolane structure (4). Chupp and Dahm later



employed ¹⁸O labeling and x-ray crystallography to confirm structure **3a** (Ar = 3,4-Cl₂C₆H₃; R¹ = CH₃; R² = H; 5-methyl group trans to the sulfinyl oxygen).³ At the same time Chupp